In many parts of the world stocks of legacy persistent organic pollutants (POPs) e.g. polychlorinated biphenyls (PCBs), organochlorine pesticides, brominated flame retardants, and fluorinated chemicals – are posing threats to human health and the environment. These POPs wastes must be destroyed or irreversibly transformed by sustainable methods that, to the greatest extent achievable, meet the following criteria:

- An effective destruction efficiency of 100% - taking into account all inputs and outputs (gaseous, liquid and solid);
- Complete containment of all process outputs to enable testing - and reprocessing if necessary - to ensure an effective destruction efficiency of 100%;
- No uncontrolled releases from the process.

Hazardous waste incinerators Over the past four decades, incineration

has been the principle method for the

destruction of organic pollutants. However, there are several problems associated with this technology. Ideally, incineration of organic compounds at temperatures higher than 850°C with sufficient turbulence and excess oxygen should lead to non-toxic endproducts, e.g. CO, and H,O. However, side reactions leading to toxic byproducts inevitably occur especially when incinerating waste containing halogens and PCDD/Fs precursors (especially PCBs, chlorophenols, chlorobenzenes and other chlorinated aromatic compounds) (Blumenstock et al., 2000; Huang and Buekens, 2001; Jiang et al., 1997; McKay, 2002) and similar brominated compounds lead to the formation of PBDD/Fs and PXDD/Fs (Weber and Kuch, 2003; Schüler and Jager, 2004). Increasingly stringent regulations that require more effective air pollution control and monitoring of gaseous emissions have resulted in a considerable reduction of releases to air of PCDD/ Fs but, in some circumstances, increased releases to air pollution control residues. However releases from waste handling together with unintentionally-formed products of incomplete combustion released to air and contained in residues from hazardous waste incinerators can

still lead to contamination around the incinerator plant and add to reservoirs of these by-products, depending on the technology used, facility operation conditions and the management of solid residues (Weber et al., 2008). The released toxic by-products, e.g. PCDD/Fs (known human carcinogens), can enter the food chain (Malisch et al., 1999). Several such cases are described in the scientific literature (Goovaerts et al., 2008; Holmes et al., 1994, 1998; Kim et al. 2006; Lovett et al., 1998).

A limiting factor of incineration is the relatively high cost (DOE, 1999; Haglund, 2007), especially when operating incinerators with state-of-the-art pollution control devices, monitoring and residue treatment. Costs also rise with the transport of hazardous wastes to the incineration facility. Cement kilns operate at temperatures of 1450°C or higher but only a few meet the necessary technical reguirements for the incineration of POPs, and high destruction efficiencies have not been demonstrated. Furthermore expensive blending, feeding and monitoring equipment is needed for effective operation (Rahuman et al., 2000). A comprehensive evaluation of destruction efficiencies is missing for incineration (hazardous waste incinerations as well as cement kilns) up to now (Weber, 2007). However, some tests suggest that incinerators achieve destruction efficiencies that are lower than those achieved by certain non-combustion technologies

These complications together with public opposition to incineration of hazardous and POPs wastes have led to the investigation of alternative and non-combustion destruction methods.

(Rahuman et al., 2000).

Non-combustion destruction methods

These technologies use physical and chemical processes of converting POPs wastes to less harmful substances. There are non-combustion, high-temperature technologies operating now at a commercial scale in one or more countries, where these technologies are licensed to destroy POPs stockpiles. Four of them are briefly described here. Further, a destruc-

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tion technology that operates under mild reaction conditions (room temperature and pressure) is also presented.

Gas Phase Chemical Reduction (GPCR)

This technology involves the gas-phase chemical reduction of organic compounds by hydrogen at temperatures of 850°C or greater and low pressure. Organic compounds are ultimately reduced to methane, hydrogen chloride (which is later neutralized), and minor amounts of low molecular weight hydrocarbons. It is able to treat both liquid and solid wastes of high contents of DDT, hexachlorobenzene (HCB). PCDD/Fs and PCB transformers. capacitors, and oils. Pre-treatment is needed for both solid and liquid wastes. The configuration of the methodology is modular: it can be fixed as well as transportable. All emissions and residues may be captured and reprocessed if needed. PCDD/ Fs have not been detected in the product gas from the process, but have been detected at low levels from natural gas burner used to heat the reaction vessel (UNEP, 2004). Solid residues are generated from solid waste inputs, but these should be suitable for disposal in a landfill, although UNEP (2003) indicates that traces of POPs may remain.

The destruction efficiency data provided by the Eco Logic company from facilities run in Australia and Canada report DEs of > 99.9999% for PCBs, DDT and HCB and > 99.9995% for PCDD/Fs (Eco Logic, 2002). The throughput of 150 tons per month or 1800 tons per year can be doubled due to modular design (IHPA, 2002). EPA (2005) stated that this technology was not considered to be cost-effective but more recently reported that it is currently "being modifying to improve its cost effectiveness" (EPA, 2010).

Base Catalysed Decomposition (BCD)

This US EPA developed process consists of two separate processing steps: first indirectly heated thermal desorption at temperatures of 200–400 °C is used for the decontamination of treated media, then the POPs in form of pure chemicals or concentrates obtained by the desorption are destroyed by a chemical reaction. The dehalogenation occurs when

selected chemicals including sodium hydroxide (a base) are mixed with the condensed contaminants and heated to 236 °C in a reactor. If the carrier oil does not meet the disposal criteria, it is returned to the reactor and reheated. This technology is able to remediate liquids, soils, sludge and sediments contaminated especially with PCBs and PCDD/Fs. The treated soil can be used as backfill on site. The technology is fixed or designed as mobile units (EPA, 2005; Rahuman et al., 2002; UNEP, 2004).

Compared to older facilities, today's plants are equipped with off-gas scrubbing and filtration. Emission concentrations are reported to be low and the total mass of emitted off-gas is orders of magnitude smaller than incinerators (UNEP, 2004).

A BCD reactor can process 2-12 tonnes of POPs per day and higher throughputs can be achieved by increasing the number of modules. High destruction efficiencies (4-6 nines) have been demonstrated for PCBs, OCPs and PCDD/Fs in treatability trials and routine operations (UNEP, 2004). The BCD treatment time is short, energy requierements are moderate and operation and maintenance costs are relatively low. The operation costs of a full-scale BCD system are considered to less than 25% of the incineration operating costs (Rahuman et al., 2000).

Super Critical Water Oxidation (SCWO)

Many chemical destruction technologies are based on the reaction of the organic compounds with hydroxyl radicals. In supercritical water, organic pollutants become highly water-soluble and react rapidly with added oxidants. Final decomposition products are carbon dioxide, water and mineral acids and salts. The supercritical phase of water occurs under high temperature and pressure conditions, e.g. 374°C and 24-28 MPa (EPA, 2005; Rahuman et al., 2000; Veriansyah et Kim, 2007). Effluent gases do not contain nitrogen oxides, acid gases, or particles and carry less than 10 ppm carbon monoxide (Environment Australia, 1997). SCWO is suitable for the treatment of soil, sludge and liquid wastes contaminated with e.g. PCBs and pesticides as well as hazardous

military wastes in low or high concentrations. The technology is highly transportable (IHPA, 2008; Marulanda, 2010).

Current SCWO plants use corrosion resistant materials. All emissions and residues may be captured for reprocessing, if needed (UNEP, 2004).

EPA (2005) as well as Rahuman et al.

(2000) claim no data on the destruction efficiency have been found, however, UNEP (2004) report a demonstrated high efficacy of the technology. IHPA (2008) mention a DE of six to eight 9s for very low pesticide concentrations. For waste containing maximum 20% of organic carbon, SCWO is considered to be far less costly than incineration (Boock, 1996). Due to some operation problems, the industrial status of SCWO is currently limited however, Veriansyah and Kim (2007) claim that enhanced investment into novel technologies potentially superior to incineration could remove the current technical limitations of the technology. More recently Marulanda (2010) suggests that the economics and performance of a mobile unit in South America might make SCWO an economically viable alternative to incineration in South American countries.

Sodium reduction

This transportable or fixed technology has been used widely for the in-situ removal of low to high contents of PCBs from transformer oils. The basic principle is the reduction of the PCBs with dispersed metallic sodium in mineral oil which leads to the final products of non-halogenated biphenyls, sodium chloride, petroleum based oil and water. The capacity for treating transformer oils has been reported to be 15 000 litres per day. No destruction efficiency has been reported for this technology and there is insufficient information available on the characterisation of residues. Also, as with any process which does not involved destruction of the transformer itself, there is concern about residues of PCBs in the porous materials of transformers when transformer oils are treated in situ. However, this approach has been widely used for treating PCBs for more than two decades (UNEP, 2004).

Ball milling/ Mechanochemical dehalogenation (MCD)

Ball mills are mechanochemical dehalogenation reactors for the destruction of PCBs and other organic chlorinated pollutants to their parent hydrocarbons in high yields. The reductive dehalogenation occurs in the presence of an alkali metal and a low acidic hydrogen source. It can be applied to contaminated materials as well as highly concentrated or pure chemicals regardless of their state. The pollutants are eliminated directly inside a contaminated material (Birke et al., 2004; UNEP, 2004). Although the mechanochemical degradation is performed at low temperatures, inside the milled material temperatures up to several thousand degrees Celsius occur when a grain collides at high velocity with a solid surface (Heinicke, 1984). The final product powder may require further treatment (IHPA, 2008).

Ball mills are available in different sizes

and constructions such that treatment of materials up to several tons is possible. Mechanochemical reduction is costeffective and offers also environmental benefits due to its low energy consumption. Due to mild reaction conditions and the design of a closed system, no harmful emissions to the environment are expected (Birke et al., 2004; UNEP, 2004). IHPA (2008) mentions the DE of ball milling to be 4-5 nines. Birke et al. (2004) state that PCBs in solid or liquid matrices can be destroyed (based on laboratory studies) to non-detectable levels in a period of minutes to hours. However, as there is currently very limited commercial experience with this technology, limited information exists on its emissions, efficacy, intermediate breakdown products and other important parameters under industrial operation. Main questions also remain regarding the destruction efficiency of the technology and the amount and toxicity of agents needed in the process (IHPA, 2008). When applied to pesticide-contaminated soils at a site in New Zealand, ball milling reduced pesticide levels in the soils by less than 90 percent. There were also concerns about possible PCDD/Fs releases to the air during processing as well as potentially

problematic levels of process agents that remained in the treated soil (PCE, 2010; PCE, 2008).

Questions to be answered

One important criterion for the assessment of POPs destruction technologies is the potential formation of new POPs and other toxic by-products during the process. PCDD/Fs can be formed during combustion of hazardous wastes (see above). However, the operation conditions for a number of non-combusiton technologies have also the potential to generate PCDD/Fs if relevant precursors are present (Weber, 2007). One example is GPCR, where if either the product gas or the ambient air used for the combustion contains chlorinated species, PCDD/ Fs may be generated. In order to meet the fundamental technical criteria for POPs destruction, both the product gas and combustion air must be suitably treated (Rahuman et al., 2000). Also the BCD treatment process may result in an increased concentration of lower chlorinated species, where this is of potential concern for PCDD/Fs, where the lower congeners are significantly more toxic than the higher congeners. It is therefore essential that the process is appropriately monitored to ensure that the reaction continues to completion (Rahuman et al., 2000). A laboratory study of PCBs destruction by SCWO has shown that in lower temperature operations considerable amounts of PCDD/Fs can be formed (Weber, 2004). Operators must ensure that processes are operated at conditions (particularly temperatue and residence times) which avoid such formation. In many cases, however, detailed assessments of non-combustion technologies with respect to PCDD/Fs formation are lacking (Weber, 2004, 2007).

How to choose the most suitable technology?

There is likely no "perffect" technology for the destruction of POPs. Many criteria play a role when deciding on a suitable technology to be applied in each specific case. A basic performance standard for POPs destruction technologies is the efficiency of destruction. Destruction

efficiency (DE) is a comparison of the amount of a specific POP that fed into a process with the total amount of that POP that is released in all the process outputs, e.g. gaseous and liquid emissions as well as solid residues. However, another measure that is sometimes used is destruction and removal efficiency (DRE), which takes into account only the amount of a specific POP that is released in air emissions. Unfortunately, many of the reports on a technology's efficiency report only DREs and sometimes DREs are erroneously reported as DEs (Costner, 2004). Both measures must be understood with respect to non detectable POPs concentrations. These are reported as "less than" detection limit values, not as zero. As a consequence, calculated DE and DRE values approach, but never reach 100 %. Therefore, a destruction efficiency of > 99.9999 % can be considered as effectively 100 % (Rahuman et al., 2000). Although chemical and toxicological analyses of all the outflow streams are expensive, they must be carried out with a frequency sufficient to ensure compliance of the technology with the basic criterion of DE > 99.9999 % during all operation conditions, i.e. startups, shutdowns, routine operation and eventual critical events (Rahuman et al., 2000). Weber (2007) states, that DEs have to be determined by long-term monitoring lasting up to months and should be performed for the whole duration of POPs destruction projects. The costs of a technology are, of course, important limiting factors. They vary significantly due to the variation in the POPs content of the material to be treated. the volume of this material and transport distances (when applying ex situ technologies). Initial investment costs can be high, however, the treatment costs of the recently developed and less optimized technologies can decrease in future which is not likely for the mature approach of incineration (Haglund, 2007). Other economical, social, environmental and technical criteria should be considered when deciding on a destruction technology, too. These are, among others (Environment Australia, 1997; IHPA, 2008; Lodolo,

2002; Rahuman et al., 2000; Veriansyah

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and Kim, 2007; Weber, 2007):

- Capability of the technology to treat the waste in the physical form(s) present (liquid, solid, size of grains), with the determined organic carbon content and necessity for the pretreatment of the waste
- Capacity of the installation
- Local applicability
- Ability of the waste to be transported and mobility of the facilities
- Reliability and maintenance of the facility
- Volumes of secondary waste streams and the content of toxic reaction byproducts under all operation conditions including unstable periods, closed system design
- Risks connected to the technology's operation (load flexibility, transient control, emergency management, dismantling of the facility)
- Public acceptability

When considering green chemistry principles (which makes the use of environmentally conscience manufacturing and production technologies) processes at room temperature and atmospheric pressures should be preferred. This ensures the lowering of energy consumption and increasing the prospects for the facility to be transported. Further, any reagents or starting materials should be inexpensive, non-toxic and easily stored, toxic by-products should be avoided and last, but not least, the method should exhibit a high destruction efficiency (Laine and Cheng, 2007). This underlines the need for the design of new reactors and further research.

Conclusions

Although the incineration of old POPs stocks is still the most used destruction technology, it cannot be considered "clean", there is concern about its high costs, comprehensive information on its destruction efficiency is missing and it is not transportable. A technology that is transportable, locally applicable, sustainable and destroys obsolete POPs at a reasonable cost would mean a breakthrough in the destruction of these chemicals (IHPA, 2008). Innovative non-combustion technologies have the potential to be-

come such technologies, however, specific conditions have to be guaranteed: long-term resource availability, strong environmental awareness and sound scientific research sustained by political support and appropriate funding to ensure continuous technological progress. Also, as some necessary information (e.g. potential PCDD/Fs formation during the destruction process) is currently provided almost exclusively by the developing and selling companies (Weber, 2007), independent evaluations of the technologies should be conducted and data made publicly available. The feasibility and durability of the technologies should be proven by real field demonstration tests in countries/regions with limited infrastructure and lack of necessary conventional supply services (IHPA, 2008).

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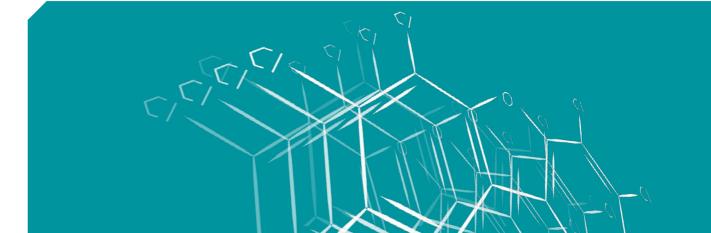
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Ministry of Foreign Affairs of the Czech Republic



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The European Union is made up of 27 Member

"The European Union is made of 27 Member States who have decided to gradually link together their know-how, resources and destinies. Together, during a period of enlargement of 50 years, they have built a zone of stability, democracy and sustainable development whilst maintaining cultural diversity, tolerance and individual freedoms. The European Union is committed to sharing its achievements and its values with countries and peoples beyond its borders".

The European Commission is the EU's executive body.

Few words about Arnika/AWHHE joint project

Among other problems related to chemical safety in Armenia is joint Czech – Armenian project "Scaling Up Experience in Improvement of Chemical Safety to Contribute to Poverty Reduction in Rural Armenia" focused on reduction of health risks related to toxic wastes at obsolete pesticides stockpiles.

The problem of Persistent Organic Pollutants (POPs) has existed in Armenia since 1980es, when annually 450-500 tons of pesticides were used in agriculture. In single years the load of pesticides application exceeded the average level of the former Soviet Union by 20 times and made up 9 kg/ha. The most vulner-

able territories in term of pesticides load were the following provinces: Ararat (Artashat, Vedi, Masis), Armavir (Armavir, Etchmiadzin), Kotayk (Kotayk, Nairi, Hrazdan), Tavush (Ijevan, Tavush, Noyemberyan), Aragatzotn (Ashtarak), Syunik (Kapan, Meghri) and Vayotz Dzor (Vayk, Yeghegnadzor).

The problem of POPs in Armenia is connected with wide application of chlorine organic compounds, as well as with accumulation of obsolete pesticides, the major part of which was either left in warehouses or buried in an unorganized manner. In 1982 was built the only burial site of obsolete pesticides where more

than 500 tons of obsolete pesticides of different groups were buried among which 60 % were persistent compounds: DDT, HCH, etc. Lack of information on toxicity of pesticides and their hazard for health and the environment is explained by the fact that after the collapse of the former Soviet Union the villager remained alone with his problems. At present there is no any coordinated mechanism and also appropriate information materials at local level. With regards to the existing materials, they are not available.

For more information look at: http://www.awhhe.am http://english.arnika.org/armenia

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